

Safety Data Sheet (SDS)

OSHA HazCom Standard 29 CFR 1910.1200(g) revised in 2012 and GHS Rev 03.

Issue date 09/13/2022

Reviewed on 09/13/2022

1 Identification

- **Product Identifier**
- **Trade Name: Carbon Steel Electrodes for Flux Cored Arc Welding**
- **Product Number:**
 - Specification: A5.20
 - Classification: E70T-1C/9C, E71T-12C/12M, E71T-11, E71T-12C-J/12M-J, E71T1-1C, E71T-GS
 - Carbon steel electrodes for flux cored arc welding
- **Relevant identified uses of the substance or mixture and uses advised against:**
For professional use only. Use according to manufacturer's specification.
- **Product Description:** Carbon steel electrodes for flux cored arc welding.
- **Application of the substance / the mixture:** Industry specific application.
- **Details of the Supplier of the Safety Data Sheet:**
- **Manufacturer/Supplier:**
Pinnacle Alloys I, LLC
9384 Wallisville Road
Houston, TX 77013
Telephone: 800-856-9353
- **Emergency telephone number:** 713-688-9353

2 Hazard(s) Identification

- **Classification of the substance or mixture:**



Health hazard

Carcinogenicity 1A

H350

May cause cancer. Route of exposure: Inhalation.

Specific Target Organ Toxicity - Repeated Exposure 1 H372-H373

Causes damage to organs through prolonged or repeated exposure. May cause damage to the respiratory system through prolonged or repeated exposure. Route of exposure: Inhalation.



Corrosion

Eye Damage 1

H318

Causes serious eye damage.



Skin Irritation 2

H315

Causes skin irritation.

Sensitization - Skin 1

H317

May cause an allergic skin reaction.

Specific Target Organ Toxicity - Single Exposure 3

H335

May cause respiratory irritation.

- **Label elements:**
- **Hazard pictograms:**



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· **Signal word:** Danger

· **Hazard-determining components of labeling:**

Iron
Lithium
Titanium Dioxide
Nickel
Titanium

· **Hazard statements:**

H315 Causes skin irritation.
H318 Causes serious eye damage.
H317 May cause an allergic skin reaction.
H350 May cause cancer. Route of exposure: Inhalation.
H335 May cause respiratory irritation.
H372-H373 Causes damage to organs through prolonged or repeated exposure. May cause damage to the respiratory system through prolonged or repeated exposure. Route of exposure: Inhalation.

· **Precautionary statements:**

P201 Obtain special instructions before use.
P202 Do not handle until all safety precautions have been read and understood.
P260 Do not breathe dust/fume/gas/mist/vapors/spray.
P264 Wash thoroughly after handling.
P270 Do not eat, drink or smoke when using this product.
P271 Use only outdoors or in a well-ventilated area.
P272 Contaminated work clothing must not be allowed out of the workplace.
P280 Wear protective gloves/protective clothing/eye protection/face protection.
P302+P352 If on skin: Wash with plenty of water.
P304+P312 IF INHALED: Call a POISON CENTER/doctor if you feel unwell.
P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P305+P351+P338 If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P308+P313 IF exposed or concerned: Get medical advice/attention.
P321 Specific treatment (see supplementary first aid instructions on this Safety Data Sheet).
P362+P364 Take off contaminated clothing and wash it before reuse.
P333+P313 If skin irritation or rash occurs: Get medical advice/attention.
P403+P233 Store in a well-ventilated place. Keep container tightly closed.
P405 Store locked up.
P501 Dispose of contents/container in accordance with local/regional/national/international regulations.

· **Unknown acute toxicity:**

This value refers to knowledge of known, established toxicological or ecotoxicological values.
10.5 % of the mixture consists of component(s) of unknown toxicity.

· **Hazard description:**

Lithium may explode when in contact with water. Exposure to moist air may result in fire. Lithium can react with water to produce flammable hydrogen gas, which may create a fire and explosion hazard. Spontaneous ignition can occur if Lithium is heated to its melting point. Lithium dusts may ignite spontaneously in moist air. Lithium can react with moisture to produce corrosive compounds. NEVER purge open drums with nitrogen before resealing. Store and transport under argon or mineral oil.

· **Classification system:** NFPA/HMIS Definitions: 0-Least, 1-Slight, 2-Moderate, 3-High, 4-Extreme

· **NFPA ratings (scale 0 - 4)**



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· HMIS-ratings (scale 0 - 4)

HEALTH	3	Health = *3
FIRE	0	Fire = 0
REACTIVITY	0	Physical Hazard = 0

· Hazard(s) not otherwise classified (HNOC): None known

* 3 Composition/Information on Ingredients

· Chemical characterization: Substance

· **Description:** Mixture of substances listed below with non-hazardous additions.

· Dangerous Components:

CAS: 7439-89-6 RTECS: NO 4565500	Iron ⚠ Flammable Solids 2, H228; ⚠ Skin Irritation 2, H315; Specific Target Organ Toxicity - Single Exposure 3, H335; Eye Irritation 2B, H320; Combustible Dust	70-98%
CAS: 1317-61-9	Iron Oxide	0-12%
CAS: 13463-67-7	Titanium Dioxide ⚠ Carcinogenicity 2, H351	0-12%
CAS: 7439-93-2 RTECS: OJ 5540000	Lithium ⚠ Substances and mixtures which, in contact with water, emit flammable gases 1, H260; ⚠ Skin Corrosion 1B, H314	0-9%
CAS: 7429-90-5 RTECS: BD 0330000	Aluminium ⚠ Flammable Solids 2, H228	0-5%
CAS: 7439-96-5 RTECS: OO 9275000	Manganese ⚠ Pyrophoric Solids 1, H250; Substances and mixtures which, in contact with water, emit flammable gases 1, H260	0-4%
CAS: 513-77-9 RTECS: CQ 8600000	Barium carbonate ⚠ Acute Toxicity - Oral 4, H302	0-3%
CAS: 7439-95-4 RTECS: OM 2100000	Magnesium ⚠ Pyrophoric Solids 1, H250; Substances and mixtures which, in contact with water, emit flammable gases 1, H260	0-3%
CAS: 7440-02-0	Nickel ⚠ Carcinogenicity 2, H351; Specific Target Organ Toxicity - Repeated Exposure 1, H372; ⚠ Sensitization - Skin 1, H317; Aquatic Acute 3, H402	0-3%
CAS: 7439-98-7 RTECS: QA 4680000	Molybdenum	0-1.2%
CAS: 1309-48-4	Magnesium Oxide ⚠ Acute Toxicity - Oral 4, H302	0-2%
CAS: 1317-95-9	Silica ⚠ Carcinogenicity 1A, H350; ⚠ Specific Target Organ Toxicity - Single Exposure 3, H335	0-3%
CAS: 1344-28-1 RTECS: BD 1200000	Aluminum Oxide ⚠ Specific Target Organ Toxicity - Single Exposure 3, H335	0-2%
CAS: 7440-21-3	Silicon ⚠ Flammable Solids 2, H228; ⚠ Acute Toxicity - Oral 4, H302; Eye Irritation 2B, H320; Combustible Dust	0-3%

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CAS: 7440-67-7 RTECS: ZH 7070000	Zirconium ⚠ Pyrophoric Solids 1, H250; Substances and mixtures which, in contact with water, emit flammable gases 1, H260	0-1%
CAS: 7631-86-9	Silicon Dioxide ⚠ Skin Irritation 2, H315; Specific Target Organ Toxicity - Single Exposure 3, H335; Eye Irritation 2B, H320	0-2%
CAS: 7440-32-6 RTECS: XR 1700000	Titanium ⚠ Skin Irritation 2, H315; Sensitization - Skin 1, H317; Eye Irritation 2B, H320	0-0.5%

· **Additional information:**

The exact percentages of the ingredients of this mixture are considered to be proprietary and are withheld in accordance with the provisions of paragraph (i) of §1910.1200 of 29 CFR 1910.1200 Trade Secrets.

Note: Certain chemical constituents listed in Section 3 may vary depending upon the Classification of the Carbon Steel Electrodes for Flux Cored Arc Welding products.

* 4 First-Aid Measures

· **Description of first aid measures**

· **General information:**

Symptoms may even occur after several hours; therefore observe area effected for at least 48 hours after the accident.

· **After inhalation:** In case of unconsciousness place patient stably in the side position for transportation.

· **After skin contact:**

Immediately wash with water and soap and rinse thoroughly.

If skin irritation occurs, consult a doctor.

Wash with soap and water.

· **After eye contact:**

Do NOT rub eyes. Immediately rinse opened eye(s) for at least 15 minutes under running water, lifting upper and lower lids occasionally. If symptoms persist, consult a physician.

If easy to do so, remove contact lenses if worn.

If eye irritation occurs, consult a doctor.

If solid or molten Lithium enters the eyes, open victim's eyes while under gently running water. Use sufficient force to open eyelids. Have victim "roll" eyes. Minimum flushing is for 30 minutes.

Seek immediate medical attention.

· **After swallowing:**

Rinse out mouth and then drink plenty of water.

Do not induce vomiting without medical advice.

If Lithium is swallowed, CALL PHYSICIAN OR POISON CONTROL CENTER FOR MOST CURRENT INFORMATION.

Do not induce vomiting. Never induce vomiting or give diluents (milk or water) to someone who is unconscious, having convulsions, or unable to swallow.

Seek immediate medical attention

· **Information for doctor**

· **Most important symptoms and effects, both acute and delayed:** Burns to eyes and skin.

· **Indication of any immediate medical attention and special treatment needed:**

Immediate medical treatment is advised in the case of eye contact or ingestion.

This material is corrosive and water reactive.

Wear chemical resistant gloves when treating victims of Lithium contamination.

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* 5 Fire-Fighting Measures

- **Extinguishing media**

- **Suitable extinguishing agents:**

Lithium is water-reactive; it will also react with carbon dioxide.

DO NOT USE WATER, CARBON DIOXIDE OR SAND.

Use DRY graphite, soda ash, powdered sodium chloride, lithium chloride, or Lith-X.

- **For safety reasons unsuitable extinguishing agents:**

DO NOT USE WATER, CARBON DIOXIDE OR SAND.

- **Special hazards arising from the substance or mixture:**

Amorphous or crystalline silicon both react exothermically when heated with alkali-metal carbonates attaining incandescence and evolving carbon monoxide. Mixtures of silicon, aluminum, and lead explode when heated. If incinerated, product will release the following toxic fumes: Oxides of silicon, aluminum, magnesium, manganese, iron, copper, molybdenum, carbon, titanium, nickel, niobium, barium, lithium, and zirconium, and fluorides and ozone.

Amorphous or crystalline silicon both react exothermically when heated with alkali-metal carbonates attaining incandescence and evolving carbon monoxide.

Material in powder form, capable of creating a dust explosion. Mixture of silicon, aluminum, and lead oxide explodes when heated.

Material in powder form is capable of creating a dust explosion. Mixture of silicon, aluminum, and lead oxide explodes when heated.

If heated to its melting point, spontaneous ignition is likely.

Lithium fires burn very hot and are difficult to extinguish. Flammable hydrogen gas and corrosive fumes are produced upon contact with water. Combustion of Lithium is accompanied by the emission of dense, white, opaque fumes that are toxic and may hide the base of the fire. Molten Lithium will burn in air, oxygen, nitrogen, and carbon dioxide. Molten Lithium may react violently with concrete or other materials containing moisture.

Explosion Sensitivity to Mechanical Impact: Not sensitive.

Explosion Sensitivity to Static Discharge: Not sensitive.

- **Advice for firefighters**

- **Special protective equipment for firefighters:**

As in any fire, wear self-contained breathing apparatus pressure-demand (NIOSH approved or equivalent) and full protective gear to prevent contact with skin and eyes.

For Incipient Fires: If incipient Lithium fires are to be fought, proper personal protective equipment must be worn. Personal protective equipment must include face-shields with safety goggles, head protection, gloves, body protection, and respiratory protection. A minimum of two sets of personal protective equipment shall be available to firefighters if incipient fires are to be fought. The gloves and body protection must be fire-retardant. In the event of fire, cool tanks with water spray. Be aware of a dangerous reaction with water, if the container is ruptured.

For Structural Fires: Proper personal protective equipment must be worn by structural firefighters. Proper protective clothing, respiratory protection, and adequate eye protection shall be used by all responding firefighting personnel assigned to a Lithium fire. Additional eye protection shall be worn by personnel wearing Self-Contained Breathing Apparatus protection to protect against the higher degree of emitted light during a Lithium fire. Visual protection equivalent to a No. 6 welding lens shall be used. Refer to NFPA 1500, "Standard on Fire Department Occupational Safety and Health Program" for more information.

For additional information, refer to NFPA 484 Standard for Combustible Metals, Metal Powders, and Metal Dusts, most recent Edition

- **Additional information:**

At temperatures above 200°C Zirconium reacts exothermically with the following: fluorine, chloride, bromide, iodine, halocarbons, carbon tetrachloride, carbon, tetra fluoride and Freon's.

Lithium fire residues shall be protected to prevent adverse reactions and to prevent the formation of reactive and unstable compounds. Lithium fire residues shall be disposed of in accordance with Federal, State, and local regulations.

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These items are not reactive, flammable, or explosive and essentially not hazardous at ambient temperatures. Welding arcs and sparks can ignite combustibles and flammable products. If involved in a fire, these products may generate irritating aluminum fumes and a variety of metal oxides. Emergency responders must wear personal protection equipment suitable for the situation. Use the extinguishing media recommended for the burning materials and fire situation. See ANSI Z49.1 "Safety in Welding and Cutting" and "Safe Practices" Code: SP, published by the American Welding Society.

* 6 Accidental Release Measures

· **Personal precautions, protective equipment and emergency procedures:**

Ensure adequate ventilation.

Avoid contact with skin, eyes and clothing.

Wear assigned protective equipment. Keep unprotected persons away.

Uncontrolled releases should be responded to by trained personnel using pre-planned procedures. Proper protective equipment should be used. In case of a spill, clear the affected area and protect people. The minimum Personal Protective Equipment recommended for response to non-incident/non-fire releases should be Level B: double-gloves (fire resistant gloves over nitrile or rubber gloves), chemical resistant suit and boots, hard-hat, and Self-Contained Breathing Apparatus.

· **Environmental precautions:** Do not allow product to reach sewage system or any water system.

· **Methods and material for containment and cleaning up:**

Dispose of contaminated material as waste according to section 13.

Ensure adequate ventilation.

Dispose of the collected material according to regulations.

Flammable solid. Stop leak if without risk. Do not touch spilled material. Use water spray curtain to divert vapor drift. Prevent entry into sewers, basements or confined areas; dike if needed. Eliminate all ignition sources.

Solid Lithium: Lithium presents an immediate safety hazard since it can react with moisture in the air and start a fire. Cover the released material quickly. Mineral oil can be used to cover Lithium. Recovered Lithium should be placed under mineral oil, in a suitable container.

Molten Lithium: This material ignites easily and reacts violently with concrete, wood, asphalt, sand, asbestos, and all gases except argon or helium. Use dry graphite or Lith-X to cover the released material and allow it to cool. Recovered Lithium should be placed under mineral oil, in a suitable container.

Spilled Lithium in Mineral Oil: Lithium covered in mineral oil is less reactive. Solid pieces should be picked up or scooped up and placed under mineral oil. If this is not possible, cover material with dry graphite. Place the absorbed material under mineral oil, in a suitable container.

· **Reference to other sections:**

See Section 7 for information on safe handling.

See Section 8 for information on personal protection equipment.

See Section 13 for disposal information.

· **PAC-1:**

7439-89-6	Iron	3.2 mg/m ³
1317-61-9	Iron Oxide	21 mg/m ³
13463-67-7	Titanium Dioxide	30 mg/m ³
7439-93-2	Lithium	3.3 mg/m ³
7439-96-5	Manganese	3 mg/m ³
513-77-9	Barium carbonate	2.2 mg/m ³
7439-95-4	Magnesium	18 mg/m ³
7440-02-0	Nickel	4.5 mg/m ³
7439-98-7	Molybdenum	30 mg/m ³
1309-48-4	Magnesium Oxide	30 mg/m ³

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1344-28-1	Aluminum Oxide	15 mg/m ³
7440-21-3	Silicon	45 mg/m ³
7440-44-0	Carbon Fiber	6 mg/m ³
7440-67-7	Zirconium	10 mg/m ³
7631-86-9	Silicon Dioxide	18 mg/m ³
7440-50-8	Copper	3 mg/m ³
554-13-2	Lithium Carbonate	3.1 mg/m ³
7440-32-6	Titanium	30 mg/m ³
7440-03-1	Niobium	30 mg/m ³
· PAC-2:		
7439-89-6	Iron	35 mg/m ³
1317-61-9	Iron Oxide	230 mg/m ³
13463-67-7	Titanium Dioxide	330 mg/m ³
7439-93-2	Lithium	36 mg/m ³
7439-96-5	Manganese	5 mg/m ³
513-77-9	Barium carbonate	270 mg/m ³
7439-95-4	Magnesium	200 mg/m ³
7440-02-0	Nickel	50 mg/m ³
7439-98-7	Molybdenum	330 mg/m ³
1309-48-4	Magnesium Oxide	120 mg/m ³
1344-28-1	Aluminum Oxide	170 mg/m ³
7440-21-3	Silicon	100 mg/m ³
7440-44-0	Carbon Fiber	330 mg/m ³
7440-67-7	Zirconium	83 mg/m ³
7631-86-9	Silicon Dioxide	740 mg/m ³
7440-50-8	Copper	33 mg/m ³
554-13-2	Lithium Carbonate	34 mg/m ³
7440-32-6	Titanium	330 mg/m ³
7440-03-1	Niobium	330 mg/m ³
· PAC-3:		
7439-89-6	Iron	150 mg/m ³
1317-61-9	Iron Oxide	1,400 mg/m ³
13463-67-7	Titanium Dioxide	2,000 mg/m ³
7439-93-2	Lithium	220 mg/m ³
7439-96-5	Manganese	1,800 mg/m ³
513-77-9	Barium carbonate	1,600 mg/m ³
7439-95-4	Magnesium	1,200 mg/m ³
7440-02-0	Nickel	99 mg/m ³
7439-98-7	Molybdenum	2,000 mg/m ³
1309-48-4	Magnesium Oxide	730 mg/m ³
1344-28-1	Aluminum Oxide	990 mg/m ³
7440-21-3	Silicon	630 mg/m ³

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7440-44-0	Carbon Fiber	2,000 mg/m ³
7440-67-7	Zirconium	500 mg/m ³
7631-86-9	Silicon Dioxide	4,500 mg/m ³
7440-50-8	Copper	200 mg/m ³
554-13-2	Lithium Carbonate	210 mg/m ³
7440-32-6	Titanium	2,000 mg/m ³
7440-03-1	Niobium	2,000 mg/m ³

* 7 Handling and Storage

· **Handling**

· **Precautions for safe handling:**

Avoid creating and breathing dust/fume/gas/mist/vapors/spray.

Ensure good ventilation/exhaustion at the workplace.

Avoid getting Lithium ON YOU or IN YOU.

Wash thoroughly with soap and large amounts of water after handling Lithium.

Avoid creating and breathing airborne dusts of Lithium.

Do not eat or drink while handling Lithium.

All employees who handle Lithium should be trained to handle it safely. Employees should be trained on the information in the SDS before working with Lithium.

Keep Lithium away from sparks, flames, and other ignition sources.

Post "No Smoking" signs in the area of use.

Use the smallest possible amount of Lithium in processes and only in designated areas.

Surplus Lithium must be returned to the container and resealed as soon as possible.

Have emergency equipment/materials (e.g., dry graphite) available.

Ensure containers are properly labeled.

Keep containers closed when not in use.

· **Information about protection against explosions and fires:** No special measures required.

· **Conditions for safe storage, including any incompatibilities**

Store away from strong acids, strong bases, strong oxidizing agents and strong reducing agents.

· **Storage**

· **Requirements to be met by storerooms and receptacles:**

Store under an inert gas (e.g., helium or argon) or mineral oil. Nitrogen should not be used as the inert gas for storage. On exposure to atmospheres that are not inert to Lithium, the material may undergo an exothermic surface reaction with oxygen or humidity; this may result in a fire. Store containers in a cool, dry location, away from direct sunlight or sources of intense heat. Store away from incompatible materials (see Section 10, Stability and Reactivity). A detached, fire-resistant building is recommended for storing large quantities. Solid Lithium shall be stored only on the ground floor. There should be no basement or depression into which water or molten metal shall be permitted to flow or fall during a fire. The storage area for Lithium must be isolated from other areas so that water cannot enter by spray or drainage from automatic sprinkler systems or any other water source. Containers of Lithium should be inspected monthly by individuals who are familiar with lithium hazards and are able to recognize the potential problems associated with the hazards. Mark and store empty containers of this product properly. Empty containers may contain residual material; therefore, empty containers must be handled with care.

For additional information, refer to NFPA 484 Standard for Combustible Metals, Metal Powders, and Metal Dusts most recent Edition. Local Fire Departments should be notified of the storage of Lithium Metal on site.

Storage and processing areas of Lithium Metal should be identified with a NFPA 704 placard (diamond) large enough to be seen from a distance.

· **Information about storage in one common storage facility:**

The storage area for Lithium must be isolated from other areas so that water cannot enter by spray or drainage from automatic sprinkler systems or any other water source.

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- **Further information about storage conditions:** Keep receptacle tightly sealed.
- **Specific end use(s):** No further relevant information available.

* 8 Exposure Controls/Personal Protection

- **Additional information about design of technical systems:**

VENTILATION AND ENGINEERING CONTROLS: Use with adequate ventilation. Mechanical exhaust may be needed. Because of the potential for reaction with water or moist air, stringent control measures such as isolation or enclosure of operations involving this material may be necessary.

- **Control parameters:**

All ventilation should be designed in accordance with OSHA standard (29 CFR 1910.94). Use local exhaust at filling zones and where leakage and dust formation is probable. Use mechanical (general) ventilation for storage areas. Use appropriate ventilation as required to keep Exposure Limits in Air below TLV & PEL limits.

- **Components with occupational exposure limits:**

The following constituents are the only constituents of the product which have a PEL, TLV or other recommended exposure limit.

At this time, the other constituents have no known exposure limits.

1317-61-9 Iron Oxide	
PEL	Short-term value: 80 mg/m ³
TWA	Long-term value: 6
13463-67-7 Titanium Dioxide	
PEL	Long-term value: 15* mg/m ³ *total dust
REL	See Pocket Guide App. A
TLV	Long-term value: 0.2* 2.5** mg/m ³ resp. fraction, *nanoscale, **finescale, A3
7429-90-5 Aluminium	
PEL	Short-term value: 5** mg/m ³ Long-term value: 15* mg/m ³ *Total dust; ** Respirable fraction
REL	Short-term value: 5** mg/m ³ Long-term value: 10* mg/m ³ as Al*Total dust**Respirable/pyro powd./welding f.
TLV	Long-term value: 1* mg/m ³ as Al; *as respirable fraction, A4
7439-96-5 Manganese	
PEL	Ceiling limit value: 5 mg/m ³ as Mn
REL	Short-term value: 3 mg/m ³ Long-term value: 1 mg/m ³ fume, as Mn
TLV	Long-term value: 0.02* 0.1** mg/m ³ as Mn; A4, *respirable **inhalable fraction
513-77-9 Barium carbonate	
PEL	Long-term value: 0.5 mg/m ³ as Ba

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REL	Long-term value: 0.5 mg/m ³ as Ba
TLV	Long-term value: 0.5 mg/m ³ as Ba, A4
7440-02-0 Nickel	
REL	Long-term value: 1 mg/m ³
REL	Long-term value: 0.015 mg/m ³ as Ni; See Pocket Guide App. A
TLV	Long-term value: 1.5* mg/m ³ elemental, *inhalable fraction, A5, BEI
7439-98-7 Molybdenum	
REL	Long-term value: 15* mg/m ³ *Total dust, as Mo
TLV	Long-term value: 10* 3** mg/m ³ as Mo; *inhalable fraction ** respirable fraction
1309-48-4 Magnesium Oxide	
REL	Long-term value: 15* mg/m ³ fume; *total particulate
TLV	Long-term value: 10* mg/m ³ *as inhalable fraction, A4
1317-95-9 Silica	
REL	Long-term value: 0.05* mg/m ³ *resp. dust; 30mg/m ³ /%SiO ₂ +2
REL	Long-term value: 0.05* mg/m ³ *respirable dust; See Pocket Guide App. A
TLV	Long-term value: 0.025* mg/m ³ *respirable particulate matter, A2
1344-28-1 Aluminum Oxide	
REL	Long-term value: 15*; 5** mg/m ³ *Total dust; ** Respirable fraction
REL	Long-term value: 10* 5** mg/m ³ as Al*Total dust**Respirable/pyro powd./welding f.
TLV	Long-term value: 1* mg/m ³ as Al; *as respirable fraction, A4
7440-21-3 Silicon	
REL	Long-term value: 15* 5** mg/m ³ *total dust **respirable fraction
REL	Long-term value: 10* 5** mg/m ³ *total dust **respirable fraction
TLV	TLV withdrawn
7440-67-7 Zirconium	
REL	Long-term value: 5 mg/m ³ as Zr
REL	Short-term value: 10 mg/m ³ Long-term value: 5 mg/m ³ as Zr

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TLV	Short-term value: 10 mg/m ³ Long-term value: 5 mg/m ³ as Zr; A4
7631-86-9 Silicon Dioxide	
ACGH	Short-term value: 3 mg/m ³ Long-term value: 10 mg/m ³
IDLH	Short-term value: 3000 mg/m ³ Long-term value: 4 mg/m ³ IDLH: Immediately dangerous to life or health
TWA	Short-term value: 6 mg/m ³ Long-term value: 4 mg/m ³
· Ingredients with biological limit values:	
7440-02-0 Nickel	
BEI	5 µg/L urine post-shift at end of workweek Nickel (background)
	30 µg/L urine post-shift at end of workweek Nickel (background)

· **Additional information:** The lists that were valid during the creation of this SDS were used as basis.

· **Exposure controls:**

· **Personal protective equipment**

· **General protective and hygienic measures:**

Keep away from foodstuffs, beverages and feed.

Immediately remove all soiled and contaminated clothing and wash before reuse.

Wash hands before breaks and at the end of work.

Avoid contact with the eyes and skin.

PROTECTION FROM MOLTEN LITHIUM: Molten Lithium shall be contained in closed systems that prevent its contact with air or reactive materials. When working with molten Lithium, the following protective equipment is recommended: flame-resistant personal protective equipment, foundry-type safety boots, fire retardant gloves, and a flame-resistant bib.

Emergency eye-wash/safety showers: where there is any possibility that an employee's eyes may be exposed to this substance, the employer should provide an eye-wash fountain/safety shower within the immediate work area for emergency use.

LOCATION OF THE EYE-WASH FOUNTAIN/ SAFETY SHOWER MUST BE SELECTED TO AVOID CONTACT OF WATER WITH THE WATER REACTIVE MATERIALS.

· **Breathing equipment:**



Suitable respiratory protective device recommended.

Respiratory protection is not generally needed when using Lithium. Maintain airborne contaminant concentrations as low as possible. If respiratory protection is needed, use only protection authorized in the U.S. Federal OSHA Standard (29 CFR 1910.134), applicable U.S. State regulations, or the Canadian CSA Standard Z94.4-93. Oxygen levels below 19.5% are considered IDLH by OSHA. In such atmospheres, use of a full-facepiece pressure/demand SCBA or a full facepiece, supplied air respirator with auxiliary self-contained air supply is required under OSHA's Respiratory Protection Standard (1910.134-1998).

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Use NIOSH approved or equivalent fume respirator or air supplied respirator when welding, brazing, cutting, grinding, or soldering in a confined space or general work area where local exhaust and/or ventilation does not keep exposure below the limits outlined in Section 8. Monitor the air quality inside the welder's helmet, and/or worker's breathing zone to determine if a respirator is required and the type needed.

· **Protection of hands:**

Due to missing tests no recommendation to the glove material can be given for the product/ the preparation/ the chemical mixture.

Select glove material based on penetration times, rates of diffusion and degradation.



Protective gloves

For Solid Lithium: Wear chemical resistant gloves for routine industrial use. Gloves should be worn in a way that they may be quickly removed in the event they become contaminated with material that ignites.

If necessary, refer to U.S. OSHA 29 CFR 1910.138 and appropriate Standards of Canada.

· **Material of gloves:**

The selection of the suitable gloves does not only depend on the material, but also on further marks of quality and varies from manufacturer to manufacturer. As the product is a preparation of several substances, the resistance of the glove material cannot be calculated in advance and has therefore to be checked prior to the application.

· **Penetration time of glove material:**

The exact break-through time has to be determined and observed by the manufacturer of the protective gloves.

· **Eye protection:**



Helmet or face shield

Wear a helmet or face shield with a filter lens around shade number 14. Adjust if needed by selecting the next lighter or darker shade number. See ANSI/ASC Z49.1 Section 4.2 or publication F2.2. Shield other workers by providing screens and flash goggles.

· **Body protection:**

For Solid Lithium: Use body protection appropriate for task taking the flammability and reactivity of the product into consideration. In most cases, protection from the potential fire hazard is of greater concern than the need for chemically impervious clothing. Chemical resistant clothing may not be fire retardant and might char or burn in contact with this product. If chemical resistant clothing is needed, fire retardant garments should be worn over the chemical resistant clothes. If a hazard of injury to the feet exists due to falling objects, rolling objects, where objects may pierce the soles of the feet or where employee's feet may be exposed to electrical hazards, wear foot protection, as described in U.S. OSHA 29 CFR 1910.136.

Wear approved head, hand, and body protection, which help to prevent injury from radiation, sparks, and electrical shock. This would include wearing welder's gloves and a protective face shield and may include arm protectors, apron, hats, shoulder protection, as well as dark, non-synthetic, substantial clothing. See ANSI Z49.1. Welders should be trained not to allow electrically live parts to contact the skin or wet clothing and gloves. The welders should insulate themselves from the work and ground and should not touch live electrical parts. Welders should not wear short sleeve shirts or short pants.

· **Limitation and supervision of exposure into the environment:** None

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Trade Name: Carbon Steel Electrodes for Flux Cored Arc Welding*** 9 Physical and Chemical Properties**· **Information on basic physical and chemical properties**· **General Information**· **Appearance:****Form:**

Flux Coated Wire/Rod

Color:

Silver/gray wire covered by various colored fluxes

· **Odor:**

Odorless until used

· **Odor threshold:**

Not determined.

· **pH-value:**

Not applicable.

· **Change in condition****Melting point/Melting range:**

Not determined.

· **Flash point:**

None

· **Flammability (solid, gaseous):**

Not determined.

· **Ignition temperature:**

Not applicable

· **Decomposition temperature:**

Not determined.

· **Auto igniting:**

Product is not self-igniting.

· **Danger of explosion:**

Product does not present an explosion hazard.

· **Explosion limits:****Lower:**

Not determined.

Upper:

Not determined.

· **Vapor pressure:**

Not applicable.

· **Density:**

Not determined.

· **Relative density:**

Not determined.

· **Vapor density:**

Not applicable.

· **Evaporation rate:**

Not applicable.

· **Solubility in / Miscibility with:****Water:**

Insoluble.

· **Partition coefficient (n-octanol/water):** Not determined.· **Viscosity:****Dynamic:**

Not applicable.

Kinematic:

Not applicable.

· **Solvent content:****VOC content:**

0.00 %

Solids content:

100.0 %

· **Other information:**

No further relevant information available.

*** 10 Stability and Reactivity**· **Reactivity:**

Stable under normal conditions.

May react violently or explosively on contact with water. Will react with water or steam to product hydrogen.

Incompatible (violent reactions) with chlorine, fluorine, oxidizers, calcium, carbide, alkali carbonates, iodine pentafluoride, cobaltic fluoride, rubidium carbide, MnF₃, nitrosyl fluoride, AgF. Mixtures of cesium acetylide with silicon react vigorously on heating. Rubidium acetylide reacts vigorously with silicon on warming.

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Lithium Ingots, foil, anodes (small pieces of foil), or rods will normally not ignite spontaneously in the air at room temperature. They will react with moisture, carbon dioxide, nitrogen, and oxygen at room temperature. The reaction is exothermic and may heat the material to the point at which it will ignite.

Dusts or finely divided Lithium (powders) can ignite spontaneously in moist air at room temperature. The greater the surface area, the more potentially reactive the Lithium. Interaction to form lithium amalgam is violently exothermic and may be explosive if large pieces of lithium are used (an amalgam is an alloy of mercury and another metal).

For additional information, refer to NFPA 484 Standard for Combustible Metals, Metal Powders, and Metal Dusts, most recent Edition and NFPA 491, "Manual of Hazardous Chemical Reactions".

• **Chemical stability:** Stable under normal conditions.

• **Thermal decomposition / conditions to be avoided:** No decomposition if used according to specifications.

• **Possibility of hazardous reactions:**

May react violently or explosively on contact with water. Will react with water or steam to product hydrogen Incompatible (violent reactions) with chlorine, fluorine, oxidizers, calcium, carbide, alkali carbonates, iodine pentafluoride, cobaltic fluoride, rubidium carbide, MnF₃, nitrosyl fluoride, AgF. Mixtures of cesium acetylide with silicon react vigorously on heating. Rubidium acetylide reacts vigorously with silicon on warming.

May react with strong oxidizers generating heat.

Reacts explosively with water.

Contact with acids or strong bases may cause generation of gas.

• **Conditions to avoid:**

Exposure to water, air, heat, flames and other sources of ignition, moisture, and incompatible chemicals.

• **Incompatible materials:**

Incompatible (violent reactions) with chlorine, fluorine, oxidizers, calcium, carbide, alkali carbonates, iodine pentafluoride, cobaltic fluoride, rubidium carbide, MnF₃, nitrosyl fluoride, AgF.

Lithium is not compatible with water, halogenated hydrocarbons (e.g., chloroform, carbon tetrachloride), Teflon, carbon dioxide, strong acids, metals (e.g., mercury, iron, nickel), strong oxidizers, alcohols, nitrogen, acid chlorides, and flammable materials. Molten Lithium can attack plastics and rubbers.

Do not store near acids, metals, chlorinated solvents, water, nitrogen.

Strong acids, strong bases, strong oxidizing agents and strong reducing agents.

• **Hazardous decomposition products:**

Hydrogen gas, lithium hydroxide, lithium carbonate, and lithium oxide.

Welding fumes and gases cannot be classified simply. The composition and quantity of both are dependent upon the metal being welded, the processes and procedures followed, and the welding consumables used.

Other conditions that also influence the composition and quantity of fumes and gases to which workers may be exposed include: coatings on the metal being welded (such as paint, plating, or galvanizing), the number of welders in operation and the volume of the work area, the quality and amount of ventilation, the position of the welder's head with respect to the fume plume, and the presence of contaminants in the atmosphere (such as chlorinated hydrocarbon vapors from cleaning and degreasing procedures). When the electrode is consumed, the fume and gas decomposition products generated are different in percent and form from the ingredients listed in Section 8. Fume and gas decomposition, and not the ingredients in the electrode, are important. The concentration of a given fume or gas component may decrease or increase by many times the original concentration. Also, new compounds not in the electrodes may form. The known gases and fumes that may form during welding or cutting and their exposure limits are noted in the list in Section 11 below. Decomposition products of normal operation include those originating from the volatilization, reaction, or oxidation of the materials shown in Section 8, plus those from the base metal and coating, etc. as noted above. Chlorinated solvents may be decomposed into toxic gases such as phosgene.

It is understood, however, that the elements and/or oxides to be mentioned are virtually always present as complex oxides and not as metals (See "Characterization of Arc Welding Fume", from the American Welding Society). The elements or oxides listed Section 8 correspond to the ACGIH categories found in "Threshold Limit Values for Chemical Substances and Physical Agents" listed in Section 8. Some products will also contain: Oxides of silicon, aluminum, magnesium, manganese, iron, copper, molybdenum, carbon, titanium, nickel, niobium, barium, lithium, and zirconium, and fluorides and ozone. Some elements or compounds may exceed their PELs/TLVs before the total fumes exceed 5 mg/m³.

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*11 Toxicological Information

· **Information on toxicological effects:**

Effects of Over-Exposure: Electric arc welding may create one or more of the following health hazards:

- ARC RAYS can injure eyes and burn skin. Incidences of skin cancer have been reported.
- ELECTRIC SHOCK can kill.
- FUMES AND GASES GENERATED FROM WELDING can be dangerous to your health.
- PRIMARY ROUTES OF ENTRY are the respiratory system, eyes, skin, and/or indigestion.
- NOISE can damage hearing.

Short-term (acute) over-exposure effects:

- WELDING FUMES may result in discomfort, such as dizziness, nausea, or dryness or irritation of the nose, throat, or eyes.
- ALUMINUM OXIDE may cause irritation of the respiratory system.
- FLUORIDES, FLUORIDE COMPOUNDS may cause skin and eye burns, pulmonary edema, and bronchitis.
- IRON, IRON OXIDE have no known effects. Treat as a nuisance dust or fume.
- MAGNESIUM, MAGNESIUM OXIDE overexposure may cause metal fume fever, characterized by metallic taste, tightness of chest, and fever. Symptoms may last 24-48 hours following overexposure.
- MANGANESE, MANGANESE COMPOUNDS may cause metal fume fever, characterized by irritation of the throat, vomiting, nausea, fever, body aches, and chills. Recovery is generally complete within 48 hours of overexposure.
- MOLYBDENUM may cause irritation of the eyes, nose, and throat.
- NICKEL, NICKEL COMPOUNDS may cause metallic taste, nausea, tightness in chest, fever, and allergic reactions.
- SILICA (amorphous) dust and fumes may cause irritation of the respiratory system, skin, and eyes.
- TITANIUM DIOXIDE may cause irritation of the respiratory system.
- COPPER may cause capillary damage, headache, cold sweat, weak pulse, and kidney and liver damage, central nervous system excitation followed by depression, jaundice, convulsions, paralysis, and coma. Death may occur from shock or renal failure.

Long-term (chronic) over-exposure effects:

- WELDING FUMES in excess levels may cause bronchial asthma, lung fibrosis, pneumoconiosis, or 'siderosis.' Overexposure to air contaminants may lead to their accumulation in the lungs, a condition which may be seen as dense areas on chest x-rays. The severity of the change is proportional to the length of exposure. The changes seen are not necessarily associated with symptoms or signs of reduced lung function or disease. In addition, the changes on X-rays may be caused by non-work factors such as smoking, etc.
- ALUMINUM OXIDE may cause pulmonary fibrosis and emphysema.
- FLUORIDES may cause serious bone erosion (osteoporosis) and mottling of teeth.
- IRON, IRON OXIDE may cause siderosis or deposits of iron in the lungs, which is believed to affect pulmonary function. Lungs will clear in time when exposure to iron fumes and its compounds ceases. Iron and magnetite (Fe₃O₄) are not regarded as fibrogenic materials.
- MANGANESE, MANGANESE COMPOUNDS may cause central nervous system effects referred to as 'manganism.' Symptoms include languor, sleepiness, muscular weakness, emotional disturbances, spastic gait, and tremors. Behavioral changes and changes in handwriting may also appear. These effects are irreversible. Employees overexposed to manganese should receive regular medical examinations for early detection of manganism.
- MOLYBDENUM prolonged overexposure may result in loss of appetite, weight loss, loss of muscle coordination, difficulty in breathing, and anemia.
- NICKEL, NICKEL COMPOUNDS may lung fibrosis or pneumoconiosis. Studies of nickel refinery workers indicated a higher incidence of lung and nasal cancers.
- SILICA (respirable crystalline silica) overexposure may result in silicosis. Respirable crystalline silica is a

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known human carcinogen. SILICA (amorphous) long term overexposure may cause pneumoconiosis. Noncrystalline forms of silica (amorphous silica) are considered to have little fibrotic potential.

- TITANIUM DIOXIDE may cause pulmonary irritation and slight fibrosis.
- COPPER may cause hepatic cirrhosis, brain damage and demyelination, kidney defects, and copper deposition in the cornea as exemplified by humans with Wilson's disease. It has also been reported that copper poisoning has led to hemolytic anemia and accelerates arteriosclerosis.

· **Acute toxicity:**

· **LD/LC50 values that are relevant for classification:**

7439-89-6 Iron		
Oral	LD50	7,500 mg/kg (Rat)
1317-61-9 Iron Oxide		
Oral	LD50	>5,000 mg/kg (Rat)
13463-67-7 Titanium Dioxide		
Oral	LD50	>10,000 mg/kg (Rat)
Dermal	LD50	>10,000 mg/kg (Rabbit)
Inhalative	LC50/4 h	>6.82 mg/l (Rat)
7439-93-2 Lithium		
Inhalative	LC50/4 h	18 mg/l (Trout)
	LC50/96 hours	62.21 mg/l (Trout)
7429-90-5 Aluminium		
Oral	LD50	>2,000 mg/kg (Rat)
Inhalative	LC50/4 h	888 mg/l (Rat)
7439-96-5 Manganese		
Oral	LD50	9,000 mg/kg (Rat)
513-77-9 Barium carbonate		
Oral	LD50	418 mg/kg (Rat)
7439-98-7 Molybdenum		
Oral	LD50	>5,000 mg/kg (Rat)
Dermal	LD50	>2,000 mg/kg (Rat)
Inhalative	LC50/4 h	800 mg/l (Trout)
		>5.84 mg/l (Rat)
1309-48-4 Magnesium Oxide		
Oral	LD50	810 mg/kg (Mouse)
1344-28-1 Aluminum Oxide		
Oral	LD50	>10,000 mg/kg (Rat)
Inhalative	LC50/4 h	>2.6 mg/l (Rat)
7440-21-3 Silicon		
Oral	LD50	3,160 mg/kg (Rat)
7631-86-9 Silicon Dioxide		
Oral	LD50	10,000 mg/kg (Rat) (OECD 401)
Dermal	LD50	5,000 mg/kg (Rabbit) (OECD 402)
Inhalative	LC50/4 h	>140->2,000 mg/l (Rat) (OCED 403)
		Maximum attainable concentration, mortality does not appear.

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10,000 mg/l (Zebra fish) (OECD 203)

· **Primary irritant effect:**· **On the skin:**

Strong caustic effect on skin and mucous membranes.

Irritant to skin and mucous membranes.

May cause an allergic skin reaction.

· **On the eye:**

Strong irritant with the danger of severe eye injury.

Corrosive effect.

Causes serious eye irritation.

· **Sensitization:** Sensitization possible through skin contact.· **Subacute to chronic toxicity:**

SYMPTOMS OF OVEREXPOSURE BY ROUTE OF EXPOSURE: In terms of anticipated occupational overexposure situations for employees, the main health effect from overexposure would be irritation or burns of contaminated skin and eyes that have been contaminated with solid Lithium (which reacts with moisture to produce corrosive compounds including lithium hydroxide) or the molten metal.

CHRONIC: Prolonged or repeated skin contact may cause dermatitis (dry, red skin). Lithium poisoning may result in kidney and central nervous system effects.

TARGET ORGANS: ACUTE: Eyes, skin, mucous membranes. CHRONIC: Skin, Nervous System, Kidney, Thyroid.

· **Additional toxicological information:**

The product shows the following dangers according to internally approved calculation methods for preparations:

Irritant

· **Carcinogenic categories:**· **IARC (International Agency for Research on Cancer):**

(a) Although IARC has classified titanium dioxide as possible carcinogenic to human (2B), their summary concludes: "No significant exposure to titanium dioxide is thought to occur during the use of products which titanium dioxide is bound to other materials, such as in cosmetics or in paints."

(b) OSHA does not regulate Titanium Dioxide as a carcinogen. However, under 29 CFR 1910.1200 the SDS must convey the fact that Titanium Dioxide is a potential carcinogen to rats.

Group 1 - Carcinogenic to humans

Group 2A - Probably carcinogenic to humans

Group 2B - Possibly carcinogenic to humans

Group 3 - Not classifiable as to its carcinogenicity to humans

Group 4 - Probably not carcinogenic to humans

13463-67-7	Titanium Dioxide	2B
7440-02-0	Nickel	2B
1317-95-9	Silica	1
7631-86-9	Silicon Dioxide	3
· NTP (National Toxicology Program):		
7440-02-0	Nickel	R
· OSHA-Ca (Occupational Safety & Health Administration):		
None of the ingredients are listed.		

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*12 Ecological Information

- **Toxicity:**

· Aquatic toxicity:	
1317-61-9 Iron Oxide	
EC50	>10,000 mg/l (Activated sludge)
13463-67-7 Titanium Dioxide	
EC50	>1,000 mg/l (Water flea)
7439-93-2 Lithium	
EC50	153.44 mg/l (Green algae)
	10 mg/l (Daphnia) (with pH-adjustment)
7439-96-5 Manganese	
EC50	40 mg/l (Water flea)
7440-02-0 Nickel	
EC50	1 mg/l (Water flea)
7631-86-9 Silicon Dioxide	
EC50	>1,000 mg/l (Daphnia) (OECD 202)

- **Persistence and degradability:** No further relevant information available.

- **Other information:**

ALL WORK PRACTICES MUST BE AIMED AT ELIMINATING ENVIRONMENTAL CONTAMINATION.

Lithium will react with water, moist air, and carbon dioxide in the atmosphere to form stable lithium salts. The effects on exposed animals would primarily be irritation and chemical burns of contaminated tissue. The main effect on plants would be the increase in salinity and alkalinity of contaminated soils if large volumes of Lithium is released. Releases of large quantities of Lithium can be detrimental to an aquatic environment by altering the salinity and alkalinity of a body of water.

- **Behavior in environmental systems:**

- **Bioaccumulative potential:** No further relevant information available.

- **Mobility in soil:** No further relevant information available.

- **Additional ecological information:**

- **General notes:**

Do not allow product to reach ground water, water course or sewage system.

Danger to drinking water if even small quantities leak into the ground.

- **Results of PBT and vPvB assessment:**

- **PBT:** Not applicable.

- **vPvB:** Not applicable.

- **Other adverse effects:** No further relevant information available.

13 Disposal Considerations

- **Waste treatment methods**

- **Recommendation:**

Must not be disposed of together with household waste. Do not allow product to reach sewage system.

Observe all federal, state and local environmental regulations when disposing of this material.

- **Waste disposal key:** D003 (Characteristic/Reactivity), applicable to wastes consisting only of this product.

- **Uncleaned packaging**

- **Recommendation:** Disposal must be made according to official regulations.

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14 Transport Information

- **UN-Number:**
· **DOT, ADR/ADN, ADN, IMDG, IATA** Non-Regulated Material
- **UN proper shipping name:**
· **DOT, ADR/ADN, ADN, IMDG, IATA** Non-Regulated Material
- **Transport hazard class(es):**
· **DOT, ADR/ADN, ADN, IMDG, IATA** Non-Regulated Material
- **Class:** Non-Regulated Material
- **Packing group:**
· **DOT, ADR/ADN, IMDG, IATA** Non-Regulated Material
- **Environmental hazards:** Not applicable.
- **Special precautions for user:** Not applicable.
- **Transport in bulk according to Annex II of MARPOL73/78 and the IBC Code:** Not applicable.
- **UN "Model Regulation":** Non-Regulated Material

* 15 Regulatory Information

- **Safety, health and environmental regulations/legislation specific for the substance or mixture:**
- **SARA (Superfund Amendments and Reauthorization):**

· **Section 355 (extremely hazardous substances):**

None of the ingredients are listed.

· **Section 313 (Specific toxic chemical listings):**

7429-90-5 Aluminium

7439-96-5 Manganese

513-77-9 Barium carbonate

7440-02-0 Nickel

1344-28-1 Aluminum Oxide

7440-50-8 Copper

554-13-2 Lithium Carbonate

· **TSCA (Toxic Substances Control Act):**

7439-89-6 Iron

ACTIVE

1317-61-9 Iron Oxide

ACTIVE

13463-67-7 Titanium Dioxide

ACTIVE

7439-93-2 Lithium

ACTIVE

7429-90-5 Aluminium

ACTIVE

7439-96-5 Manganese

ACTIVE

513-77-9 Barium carbonate

ACTIVE

7439-95-4 Magnesium

ACTIVE

7440-02-0 Nickel

ACTIVE

7439-98-7 Molybdenum

ACTIVE

1309-48-4 Magnesium Oxide

ACTIVE

1344-28-1 Aluminum Oxide

ACTIVE

7440-21-3 Silicon

ACTIVE

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7440-44-0	Carbon Fiber	ACTIVE
7440-67-7	Zirconium	ACTIVE
7631-86-9	Silicon Dioxide	ACTIVE
66402-68-4	Ground Limestone	ACTIVE
7440-50-8	Copper	ACTIVE
554-13-2	Lithium Carbonate	ACTIVE
7440-32-6	Titanium	ACTIVE
7440-03-1	Niobium	ACTIVE
· Hazardous Air Pollutants		
7439-96-5	Manganese	

· California Proposition 65:

WARNING: This product can expose you to chemicals including the listed chemicals which are known to the State of California to cause cancer, birth defects and/or other reproductive harm. For more information, go to www.P65Warnings.ca.gov.

· Chemicals known to cause cancer:		
13463-67-7	Titanium Dioxide	
7440-02-0	Nickel	
· Chemicals known to cause reproductive toxicity for females:		
None of the ingredients are listed.		
· Chemicals known to cause reproductive toxicity for males:		
None of the ingredients are listed.		
· Chemicals known to cause developmental toxicity:		
554-13-2	Lithium Carbonate	
· New Jersey Right-to-Know List:		
13463-67-7	Titanium Dioxide	
7439-93-2	Lithium	
7429-90-5	Aluminium	
7439-96-5	Manganese	
7439-95-4	Magnesium	
7440-02-0	Nickel	
7439-98-7	Molybdenum	
1309-48-4	Magnesium Oxide	
1317-95-9	Silica	
1344-28-1	Aluminum Oxide	
7440-21-3	Silicon	
7440-67-7	Zirconium	
7440-50-8	Copper	
554-13-2	Lithium Carbonate	
7440-32-6	Titanium	
· New Jersey Special Hazardous Substance List:		
7439-93-2	Lithium	F2, R2

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7429-90-5	Aluminium	F3, R1
7439-96-5	Manganese	F3, R1
7440-02-0	Nickel	CA
1317-95-9	Silica	CA
7440-21-3	Silicon	F3
7440-67-7	Zirconium	F4, R1
554-13-2	Lithium Carbonate	TE
7440-32-6	Titanium	F3, R1

· **Pennsylvania Right-to-Know List:**

13463-67-7	Titanium Dioxide	
7439-93-2	Lithium	
7429-90-5	Aluminium	
7439-96-5	Manganese	
7439-95-4	Magnesium	
7440-02-0	Nickel	
7439-98-7	Molybdenum	
1309-48-4	Magnesium Oxide	
1317-95-9	Silica	
1344-28-1	Aluminum Oxide	
7440-21-3	Silicon	
7440-67-7	Zirconium	
7631-86-9	Silicon Dioxide	
7440-50-8	Copper	

· **Pennsylvania Special Hazardous Substance List:**

7429-90-5	Aluminium	E
7439-96-5	Manganese	E
7440-02-0	Nickel	ES
1344-28-1	Aluminum Oxide	E
7440-50-8	Copper	E

· **Carcinogenic categories:**

· **EPA (Environmental Protection Agency):**

7439-96-5	Manganese	D
513-77-9	Barium carbonate	D, CBD(inh), NL(oral)
7440-50-8	Copper	D

· **TLV (Threshold Limit Value established by ACGIH):**

13463-67-7	Titanium Dioxide	A4
7429-90-5	Aluminium	A4
513-77-9	Barium carbonate	A4
7440-02-0	Nickel	A5
7439-98-7	Molybdenum	A3
1309-48-4	Magnesium Oxide	A4

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Safety Data Sheet (SDS)

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1317-95-9	Silica	A2
1344-28-1	Aluminum Oxide	A4
7440-67-7	Zirconium	A4
· NIOSH-Ca (National Institute for Occupational Safety and Health):		
13463-67-7	Titanium Dioxide	
7440-02-0	Nickel	
1317-95-9	Silica	

· **GHS label elements**

The product is classified and labeled according to the Globally Harmonized System (GHS).

· **Hazard pictograms:**



· **Signal word:** Danger

· **Hazard-determining components of labeling:**

Iron
Lithium
Titanium Dioxide
Nickel
Titanium

· **Hazard statements:**

H315 Causes skin irritation.
H318 Causes serious eye damage.
H317 May cause an allergic skin reaction.
H350 May cause cancer. Route of exposure: Inhalation.
H335 May cause respiratory irritation.
H372-H373 Causes damage to organs through prolonged or repeated exposure. May cause damage to the respiratory system through prolonged or repeated exposure. Route of exposure: Inhalation.

· **Precautionary statements:**

P201 Obtain special instructions before use.
P202 Do not handle until all safety precautions have been read and understood.
P260 Do not breathe dust/fume/gas/mist/vapors/spray.
P264 Wash thoroughly after handling.
P270 Do not eat, drink or smoke when using this product.
P271 Use only outdoors or in a well-ventilated area.
P272 Contaminated work clothing must not be allowed out of the workplace.
P280 Wear protective gloves/protective clothing/eye protection/face protection.
P302+P352 If on skin: Wash with plenty of water.
P304+P312 IF INHALED: Call a POISON CENTER/doctor if you feel unwell.
P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.
P305+P351+P338 If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.
P308+P313 IF exposed or concerned: Get medical advice/attention.
P321 Specific treatment (see supplementary first aid instructions on this Safety Data Sheet).
P362+P364 Take off contaminated clothing and wash it before reuse.
P333+P313 If skin irritation or rash occurs: Get medical advice/attention.
P403+P233 Store in a well-ventilated place. Keep container tightly closed.
P405 Store locked up.

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P501 Dispose of contents/container in accordance with local/regional/national/international regulations.

· **National regulations:**

The product is not subject to be labelled according with the prevailing version of the regulations on hazardous substances.

· **Chemical safety assessment:** A Chemical Safety Assessment has not been carried out.

16 Other Information

Pinnacle Alloys urges each end user and recipient of this SDS to study it carefully. If necessary, consult an industrial hygienist or other expert to understand this information and safeguard the environment and protect workers from potential hazards associated with the handling or use of this product. This information is believed to be accurate as of the revision date shown above. However, no warranty, expressed or implied, is given. Because the conditions or methods of use are beyond Pinnacle Alloys' control, we assume no liability resulting from the use of this product. Regulatory requirements are subject to change and may differ between various locations. Compliance with all applicable Federal, State, Provincial, and Local laws and regulations remain the responsibility of the user.

· **Contact:**

· **Abbreviations and acronyms:**

ADR: The European Agreement concerning the International Carriage of Dangerous Goods by Road

ADN: The European Agreement concerning the International Carriage of Dangerous Goods by Inland Waterways

IMDG: International Maritime Code for Dangerous Goods

DOT: US Department of Transportation

IATA: International Air Transport Association

EINECS: European Inventory of Existing Commercial Chemical Substances

ELINCS: European List of Notified Chemical Substances

CAS: Chemical Abstracts Service (division of the American Chemical Society)

NFPA: National Fire Protection Association (USA)

HMIS: Hazardous Materials Identification System (USA)

VOC: Volatile Organic Compounds (USA, EU)

LC50: Lethal concentration, 50 percent

LD50: Lethal dose, 50 percent

PBT: Persistent, Bioaccumulative and Toxic

vPvB: very Persistent and very Bioaccumulative

NIOSH: National Institute for Occupational Safety and Health

OSHA: Occupational Safety & Health Administration

TLV: Threshold Limit Value

PEL: Permissible Exposure Limit

REL: Recommended Exposure Limit

BEI: Biological Exposure Limit

Flammable Solids 2: Flammable solids – Category 2

Pyrophoric Solids 1: Pyrophoric solids – Category 1

Substances and mixtures which, in contact with water, emit flammable gases 1: Substances and mixtures which in contact with water emit flammable gases – Category 1

Acute Toxicity - Oral 4: Acute toxicity – Category 4

Skin Corrosion 1B: Skin corrosion/irritation – Category 1B

Skin Irritation 2: Skin corrosion/irritation – Category 2

Eye Damage 1: Serious eye damage/eye irritation – Category 1

Eye Irritation 2B: Serious eye damage/eye irritation – Category 2B

Sensitization - Skin 1: Skin sensitisation – Category 1

Carcinogenicity 1A: Carcinogenicity – Category 1A

Carcinogenicity 2: Carcinogenicity – Category 2

Specific Target Organ Toxicity - Single Exposure 3: Specific target organ toxicity (single exposure) – Category 3

Specific Target Organ Toxicity - Repeated Exposure 1: Specific target organ toxicity (repeated exposure) – Category 1

Aquatic Acute 3: Hazardous to the aquatic environment - acute aquatic hazard – Category 3

· *** Data compared to the previous version altered.**

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